

MSDS

Version: 1.0

CreationDate: Dec. 6, 2018

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1. Identification

1.1 Product name

Acrylamide

1.2 Other means of identification

Product number F010317
Other names vinylamide

1.3 Recommended use of the chemical and restrictions on use

Identified uses Used for research and development only. Acrylamide is used as a reactive monomer and intermediate in the production of organic chemicals and in the synthesis of polyacrylamides. Acrylamide is also used as a flocculent for sewage and waste treatment, soil conditioning agents, ore processing, paper and textile industries, and in the manufacture of dyes, adhesives, and permanent press fabrics.

Uses advised against no data available

1.4 Distributor's details

Company Ecodell Co., Ltd.
Address F226, 45, Jojeong-daero, Hanam-si, Gyeonggi-do, Korea
Telephone +82-2-457-2236
Fax +82-2-6442-2236

1.5 Emergency phone number

Emergency phone number +82-2-457-2236
Service hours Monday to Friday, 9am-6pm (Standard time zone: UTC/GMT +9 hours).

2. Hazard identification

2.1 Classification of the substance or mixture

Acute toxicity - Oral, Category 3

Acute toxicity - Dermal, Category 4

Skin irritation, Category 2

Eye irritation, Category 2

Skin sensitization, Category 1

Acute toxicity - Inhalation, Category 4

Germ cell mutagenicity, Category 1B

Carcinogenicity, Category 1B

Specific target organ toxicity – repeated exposure, Category 1

Reproductive toxicity, Category 2

2.2 GHS label elements, including precautionary statements

Pictogram(s)



Signal word

Danger

Hazard statement(s)

H301 Toxic if swallowed

H312 Harmful in contact with skin

H315 Causes skin irritation

H319 Causes serious eye irritation

H317 May cause an allergic skin reaction

H332 Harmful if inhaled

H340 May cause genetic defects

H350 May cause cancer

H372 Causes damage to organs through prolonged or repeated exposure

Precautionary statement(s)

Prevention

P264 Wash ... thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P261 Avoid breathing dust/fume/gas/mist/vapours/spray.

P272 Contaminated work clothing should not be allowed out of the workplace.

P271 Use only outdoors or in a well-ventilated area.

P201 Obtain special instructions before use.

P202 Do not handle until all safety precautions have been read and understood.

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

Response

P301+P310 IF SWALLOWED: Immediately call a POISON CENTER/doctor/...

P321 Specific treatment (see ... on this label).

P330 Rinse mouth.

P302+P352 IF ON SKIN: Wash with plenty of water/...

P312 Call a POISON CENTER/doctor/...if you feel unwell.

P362+P364 Take off contaminated clothing and wash it before reuse.

P332+P313 If skin irritation occurs: Get medical advice/attention.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

P337+P313 If eye irritation persists: Get medical advice/attention.

P333+P313 If skin irritation or rash occurs: Get medical advice/attention.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P308+P313 IF exposed or concerned: Get medical advice/attention.

P314 Get medical advice/attention if you feel unwell.

Storage

P405 Store locked up.

Disposal

P501 Dispose of contents/container to ...

2.3 Other hazards which do not result in classification

no data available

3. Composition/information on ingredients

3.1 Substances

Chemical name	Common names and synonyms	CAS number	EC number	Concentration
Acrylamide	Acrylamide	79-06-1	201-173-7	≥99.9%

4. First-aid measures

4.1 Description of necessary first-aid measures

General advice

Medical attention is required. Consult a doctor. Show this safety data sheet (SDS) to the doctor in attendance.

If inhaled

Fresh air, rest. Refer for medical attention.

Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention. Wear protective gloves when administering first aid.

Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

Following ingestion

Rinse mouth. Give one or two glasses of water to drink. Refer immediately for medical attention.

4.2 Most important symptoms/effects, acute and delayed

Classified as very toxic; probable oral lethal human dose is between 50 and 500 mg/kg or between 1 teaspoon and 1 ounce for a 150 lb. person. Polymerized acrylamide is not toxic, but the monomer can cause peripheral nerve damage. It is a cumulative neurotoxin and repeated exposure to small amounts may cause serious injury to the nervous system. The neurological effects may be delayed. Polymer inhibitors or stabilizers added to the monomer may also produce toxicity. The symptoms of acrylamide toxicity are consistent with mid-brain lesions and blocked transport along both motor and sensory axons. Individuals with nervous system diseases should not be exposed to acrylamide. (EPA, 1998)

Excerpt from ERG Guide 153P [Substances - Toxic and/or Corrosive (Combustible)]: TOXIC; inhalation, ingestion or skin contact with material may cause severe injury or death. Contact with molten substance may cause severe burns to skin and eyes. Avoid any skin contact. Effects of contact or inhalation may be delayed. Fire may produce irritating, corrosive and/or toxic gases. Runoff from fire control or dilution water may be corrosive and/or toxic and cause pollution. (ERG, 2016)

Excerpt from ERG Guide 128 [Flammable Liquids (Water-Immiscible)]: Inhalation or contact with material may irritate or burn skin and eyes. Fire may produce irritating, corrosive and/or toxic gases. Vapors may cause dizziness or suffocation. Runoff from fire control or dilution water may cause pollution. (ERG, 2016)

4.3 Indication of immediate medical attention and special treatment

needed, if necessary

Absorption, Distribution and Excretion

MICROSPHERES OF (14)C-LABELED POLYACRYLAMIDE WERE MAINLY (APPROX 80%) FOUND IN LIVER & SPLEEN BOTH AFTER IV & IP INJECTION IN MOUSE & RAT, ALSO DETECTED EARLY (1 HR AFTER IV INJECTION) IN BONE MARROW, & PARTICLE AGGREGATES WERE ALSO INITIALLY FOUND IN LUNGS.

5. Fire-fighting measures

5.1 Extinguishing media

Suitable extinguishing media

Fight fire from maximum distance. Dike fire control water for later disposal; do not scatter the material. For small fires, use dry chemical, carbon dioxide, water spray or foam. For large fires use water spray, fog or foam. Move container from fire area if you can do so without risk. (EPA, 1998)

5.2 Specific hazards arising from the chemical

Pure acrylamide will decompose at 347-572F giving ammonia, hydrogen and carbon monoxide. Avoid strong oxidizers. Avoid heat, ultraviolet light. Hazardous polymerization may occur. It readily polymerizes when heated to the melting point or when exposed to ultraviolet light. It is known to polymerize with violence when heated. (EPA, 1998)

Excerpt from ERG Guide 153P [Substances - Toxic and/or Corrosive (Combustible)]: Combustible material: may burn but does not ignite readily. When heated, vapors may form explosive mixtures with air: indoors, outdoors and sewers explosion hazards. Those substances designated with a (P) may polymerize explosively when heated or involved in a fire. Contact with metals may evolve flammable hydrogen gas. Containers may explode when heated. Runoff may pollute waterways. Substance may be transported in a molten form. (ERG, 2016)

Excerpt from ERG Guide 128 [Flammable Liquids (Water-Immiscible)]: HIGHLY FLAMMABLE: Will be easily ignited by heat, sparks or flames. Vapors may form explosive mixtures with air. Vapors may travel to source of ignition and flash back. Most vapors are heavier than air. They will spread along ground and collect in low or confined areas (sewers, basements, tanks). Vapor explosion hazard indoors, outdoors or in sewers. Those substances designated with a (P) may polymerize explosively when heated or involved in a fire. Runoff to sewer may create fire or explosion hazard. Containers may explode when heated. Many liquids are lighter than water. Substance may be transported hot. For hybrid vehicles, ERG Guide 147 (lithium ion batteries) or ERG Guide 138 (sodium batteries) should also be consulted. If molten aluminum is involved, refer to ERG Guide 169. (ERG, 2016)

5.3 Special protective actions for fire-fighters

Wear self-contained breathing apparatus for firefighting if necessary.

6. Accidental release measures

6.1 Personal precautions, protective equipment and emergency procedures

Avoid dust formation. Avoid breathing mist, gas or vapours. Avoid contacting with skin and eye. Use personal protective equipment. Wear chemical impermeable gloves. Ensure adequate ventilation. Remove all sources of ignition. Evacuate personnel to safe areas. Keep people away from and upwind of spill/leak.

6.2 Environmental precautions

Consult an expert! Personal protection: chemical protection suit including self-contained breathing apparatus. Do NOT let this chemical enter the environment. Sweep spilled substance into covered sealable containers. If appropriate, moisten first to prevent dusting. Carefully collect remainder. Then store and dispose of according to local regulations.

6.3 Methods and materials for containment and cleaning up

Collect and arrange disposal. Keep the chemical in suitable and closed containers for disposal. Remove all sources of ignition. Use spark-proof tools and explosion-proof equipment. Adhered or collected material should be promptly disposed of, in accordance with appropriate laws and regulations.

7. Handling and storage

7.1 Precautions for safe handling

Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

7.2 Conditions for safe storage, including any incompatibilities

Separated from incompatible materials. See Chemical Dangers. Cool. Keep in the dark. Well closed. Store in an area without drain or sewer access.

8. Exposure controls/personal protection

8.1 Control parameters

Occupational Exposure limit values

Component	Acrylamide			
CAS No.	79-06-1			
	Limit value - Eight hours		Limit value - Short term	
	ppm	mg/m ³	ppm	mg/m ³
Australia		0,03		
Austria		0,03		0,12
Belgium		0,03		
Canada - Ontario		0,03 (1)		
Canada - Québec	0,03			
Denmark		0,03		0,06
Finland		0,03		
France	0,1	0,3		
Germany (AGS)		0,07 (1)(2)		1,2 (3)(4)
		0,15 (3)		
Hungary				0,03
Ireland		0,03		

Component	Acrylamide			
CAS No.	79-06-1			
Japan		0,1		
Japan - JSOH		0,1		
Latvia		0,2		
New Zealand		0,03		
People's Republic of China		0,3		
Poland		0,1		
Singapore		0,03		
South Korea		0,03		
Spain		0,03		
Sweden		0,03		0,1 (1)
Switzerland		0,03 inhalable aerosol		
The Netherlands		0,16		
USA - NIOSH		0,03		
USA - OSHA		0,3		
United Kingdom		0,3		
	Remarks			
Austria	TRK value (based on technical feasibility)			
Canada - Ontario	(1) Inhalable aerosol and vapour			
Germany (AGS)	(1) Workplace exposure concentration corresponding to the proposed preliminary acceptable cancer risk. (see background document: Germany AGS) (2) When using a technique that represents the state of the art the target concentration will not be exceeded. (3) Workplace exposure concentration corresponding to the proposed tolerable cancer risk. (see background document: Germany AGS) (4) 15 minutes average value			
Spain	skin, sen			
Sweden	(1) 15 minutes average value			

8.2 Appropriate engineering controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

8.3 Individual protection measures, such as personal protective equipment (PPE)

Eye/face protection

Wear tightly fitting safety goggles with side-shields conforming to EN 166(EU) or NIOSH (US).

Skin protection

Wear fire/flammable resistant and impervious clothing. Handle with gloves. Gloves must be inspected prior to use. Wash and dry hands. The selected protective gloves have to satisfy the specifications of EU Directive 89/686/EEC and the standard EN 374 derived from it.

Respiratory protection

If the exposure limits are exceeded, irritation or other symptoms are experienced, use a full-face respirator.

Thermal hazards

no data available

9. Physical and chemical properties

Physical state	Solid.
Colour	-
Odour	no data available
Melting point/ freezing point	84.5 °C. Atm. press.:1 atm.
Boiling point or initial boiling point and boiling range	125 °C
Flammability	Combustible Solid (may also be dissolved in flammable liquids).
Lower and upper explosion limit / flammability limit	no data available
Flash point	138 °C
Auto-ignition temperature	240°C (USCG, 1999)
Decomposition temperature	no data available
pH	no data available
Kinematic viscosity	no data available
Solubility	Miscible with water
Partition coefficient n-octanol/water	log Pow = -0.9. Temperature:20 °C.
Vapour pressure	0.9 Pa. Temperature:25 °C.;4.4 Pa. Temperature:40 °C.;11 Pa. Temperature:50 °C.
Density and/or relative density	1.12. Temperature:30 °C.
Relative vapour density	2.45 (vs air)
Particle characteristics	no data available

10. Stability and reactivity

10.1 Reactivity

The substance polymerizes violently due to heating above 85°C or under the influence of light and oxidants. It reacts with strong bases and strong oxidants. Decomposes on burning. This produces toxic and corrosive fumes including nitrogen oxides.

Very soluble in water.

10.2 Chemical stability

no data available

10.3 Possibility of hazardous reactions

Amides, such as ACRYLAMIDE, react with azo and diazo compounds to generate toxic gases. Flammable gases are formed by the reaction of organic amides/imides with strong reducing agents. Amides are very weak bases (weaker than water). Mixing amides with dehydrating agents such as P₂O₅ or SOCl₂ generates the corresponding nitrile. The combustion of these compounds generates mixed oxides of nitrogen (NO_x). Spontaneous, violent polymerization occurs at its melting point (86°C) [Bretherick, 5th ed., 1995, p. 428]. Can polymerize vigorously if mixed with peroxides.

10.4 Conditions to avoid

no data available

10.5 Incompatible materials

CHEMICAL PROFILE: White, crystalline solid, toxic, confirmed carcinogen, absorbed through the skin. When heated to decomposition it emits toxic fumes of oxides of nitrogen. Spontaneous, violent polymerization occurs at its melting point (86 deg. C) [Bretherick, 5th ed., 1995, p. 428]. (REACTIVITY, 1999)

10.6 Hazardous decomposition products

no data available

11. Toxicological information

Acute toxicity

- Oral: no data available
- Inhalation: no data available
- Dermal: LD50 - rabbit (male/female) - 1 141 mg/kg bw.

Skin corrosion/irritation

no data available

Serious eye damage/irritation

no data available

Respiratory or skin sensitization

no data available

Germ cell mutagenicity

no data available

Carcinogenicity

NTP: Reasonably anticipated to be a human carcinogen

Reproductive toxicity

No information is available on the reproductive or developmental effects of acrylamide in humans. In one animal study, decreases in body weight and body weight gain and an increase in preimplantation loss were observed in rats orally exposed to acrylamide. In mice orally exposed to acrylamide, decreased sperm counts were reported.

STOT-single exposure

no data available

STOT-repeated exposure

no data available

Aspiration hazard

no data available

12. Ecological information

12.1 Toxicity

- Toxicity to fish: LC50 - *Oncorhynchus mykiss* (previous name: *Salmo gairdneri*) - 180 ppm - 96 h.
- Toxicity to daphnia and other aquatic invertebrates: EC50 - *Daphnia magna* - 98 mg/L - 48 h.
- Toxicity to algae: IC50 - *Pseudokirchneriella subcapitata* (previous names: *Raphidocelis subcapitata*, *Selenastrum capricornutum*) - 67.7 mg/L - 72 h.
- Toxicity to microorganisms: IC50 - *Pseudokirchneriella subcapitata* (previous names: *Raphidocelis subcapitata*, *Selenastrum capricornutum*) - 67.7 mg/L - 72 h.

12.2 Persistence and degradability

no data available

12.3 Bioaccumulative potential

no data available

12.4 Mobility in soil

no data available

12.5 Other adverse effects

no data available

13. Disposal considerations

13.1 Disposal methods

Product

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

Contaminated packaging

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

14. Transport information

14.1 UN Number

ADR/RID: UN2074

IMDG: UN2074

IATA: UN2074

14.2 UN Proper Shipping Name

ADR/RID: ACRYLAMIDE, SOLID

IMDG: ACRYLAMIDE, SOLID

IATA: ACRYLAMIDE, SOLID

14.3 Transport hazard class(es)

ADR/RID: 6.1

IMDG: 6.1

IATA: 6.1

14.4 Packing group, if applicable

ADR/RID: III

IMDG: III

IATA: III

14.5 Environmental hazards

ADR/RID: no

IMDG: no

IATA: no

14.6 Special precautions for user

no data available

14.7 Transport in bulk according to Annex II of MARPOL 73/78 and the IBC Code

no data available

15. Regulatory information

15.1 Safety, health and environmental regulations specific for the product in question

Chemical name	Common names and synonyms	CAS number	EC number
Acrylamide	Acrylamide	79-06-1	201-173-7
European Inventory of Existing Commercial Chemical Substances (EINECS)			Listed.
EC Inventory			Listed.
United States Toxic Substances Control Act (TSCA) Inventory			Listed.
China Catalog of Hazardous chemicals 2015			Listed.
New Zealand Inventory of Chemicals (NZIoC)			Listed.
Philippines Inventory of Chemicals and Chemical Substances (PICCS)			Listed.
Vietnam National Chemical Inventory			Listed.

16. Other information

Information on revision

Creation Date Feb. 6, 2018

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Abbreviations and acronyms

- CAS: Chemical Abstracts Service
- ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road
- RID: Regulation concerning the International Carriage of Dangerous Goods by Rail
- IMDG: International Maritime Dangerous Goods
- IATA: International Air Transportation Association
- TWA: Time Weighted Average
- STEL: Short term exposure limit
- LC50: Lethal Concentration 50%
- LD50: Lethal Dose 50%
- EC50: Effective Concentration 50%

References

- IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>
- HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>
- IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>
- eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en
- CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>
- ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>
- ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>
- Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>
- ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

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